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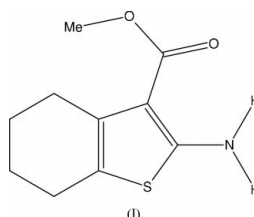
## Key indicators

Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.008\text{ \AA}$   
 $R$  factor = 0.078  
 $wR$  factor = 0.190  
Data-to-parameter ratio = 13.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## Methyl 2-amino-4,5,6,7-tetrahydro-1-benzothiophene-3-carboxylate

The molecule of the title compound,  $\text{C}_{10}\text{H}_{13}\text{NO}_2\text{S}$ , is planar. The crystal structure is stabilized by intra- and intermolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds.Received 1 April 2004  
Accepted 5 April 2004  
Online 17 April 2004

## Comment

2-Aminothiophenes are a useful class of compounds because they display different biological properties, such as anti-tubercular, bacteriostatic and antifungal activities (Nakanishi *et al.*, 1970). The structure of the title compound, (I), is composed of a thiophene ring substituted by a 2-amino group and a 3-methyl ester group. The molecule is planar (Fig. 1). The terminal ester group lies in the plane of the thiophene ring system, the torsion angle  $\text{C}10-\text{O}2-\text{C}9-\text{C}2$  being  $178.0(4)^\circ$ .The carbonyl group has a *cis* orientation with respect to the  $\text{C}2=\text{C}3$  double bond. This leads to a strong intramolecular hydrogen bond (Table 1) between one of the H atoms, *viz.*  $\text{H}2\text{N}$  of the amino group, and the O atom of the carbonyl group. The other H atom,  $\text{H}1\text{N}$ , is involved in molecular chains along the *c* axis (Fig. 2).

## Experimental

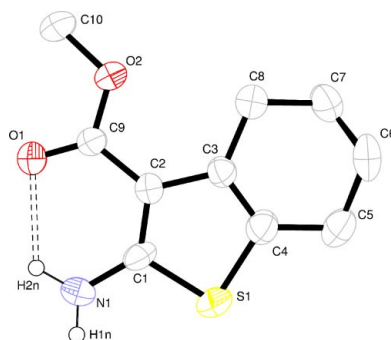
The title compound, (I), was synthesized by the Gewald reaction (Gewald *et al.*, 1966) by mixing cyclohexanone (0.04 mol), ethyl cyanoacetate (0.04 mol), sulfur (0.04 mol) and 40 ml of ethanol and stirring the mixture at 325 K for 1 h, with dropwise addition of 4 ml of dimethylamine to yield the ester, followed by alkaline hydrolysis

Figure 1

View of the molecule of (I), drawn with 50% probability displacement ellipsoids. The hydrogen bond is shown as dashed lines.

using sodium hydroxide solution. It was recrystallized from methanol and crystals of suitable size were removed for single-crystal studies.

#### Crystal data

$C_{10}H_{13}NO_2S$	$D_x = 1.363 \text{ Mg m}^{-3}$
$M_r = 211.28$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 450 reflections
$a = 16.915 (3) \text{ \AA}$	$\theta = 1.6\text{--}25.6^\circ$
$b = 14.003 (3) \text{ \AA}$	$\mu = 0.29 \text{ mm}^{-1}$
$c = 10.142 (2) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 121.018 (3)^\circ$	Plate, yellow
$V = 2058.9 (7) \text{ \AA}^3$	$0.17 \times 0.16 \times 0.07 \text{ mm}$
$Z = 8$	

#### Data collection

Bruker SMART CCD area-detector diffractometer	1808 independent reflections
$\varphi$ and $\omega$ scans	1337 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1997)	$R_{\text{int}} = 0.054$
$T_{\text{min}} = 0.918$ , $T_{\text{max}} = 0.980$	$\theta_{\text{max}} = 27.8^\circ$
1364 measured reflections	$h = -21 \rightarrow 21$
	$k = -18 \rightarrow 17$
	$l = -12 \rightarrow 12$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0929P)^2 + 1.4124P]$
$R[F^2 > 2\sigma(F^2)] = 0.078$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.190$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.18$	$\Delta\rho_{\text{max}} = 0.54 \text{ e \AA}^{-3}$
1808 reflections	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
136 parameters	
H atoms treated by a mixture of independent and constrained refinement	

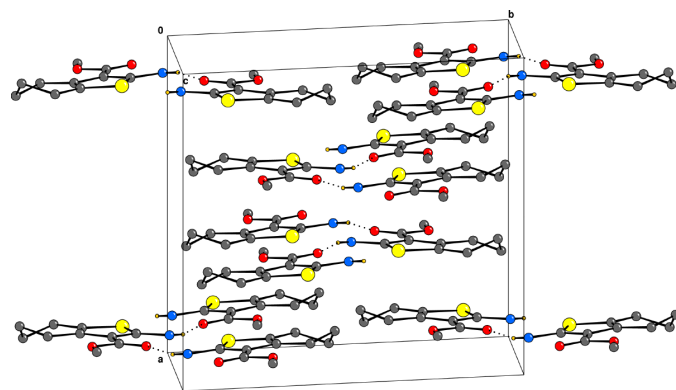
**Table 1**

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$N1\text{--}H1N\cdots O1^i$	0.84 (5)	2.17 (5)	2.972 (5)	161 (4)
$N1\text{--}H2N\cdots O1$	0.94 (6)	2.14 (6)	2.766 (6)	123 (4)

Symmetry code: (i)  $x, -y, \frac{1}{2} + z$ .

The H atoms of the amino group were located in difference Fourier maps and refined freely. The N—H bond lengths are 0.84 (5) and 0.94 (6)  $\text{\AA}$ . The methyl H atoms were constrained to an ideal geometry [ $C\text{--}H = 0.96 \text{ \AA}$  and  $U_{\text{iso}} = 1.5U_{\text{eq}}(C)$ ], but were allowed to rotate freely about the C—C bond. All the remaining H atoms were placed in idealized positions and constrained to ride on their parent atoms [ $C\text{--}H = 0.97 \text{ \AA}$  and  $U_{\text{iso}} = 1.2U_{\text{eq}}(C)$ ].



**Figure 2**

Packing diagram of (I), showing N—H $\cdots$ O hydrogen-bond chains as dashed lines.

Data collection: SMART (Bruker, 1998); cell refinement: SMART; data reduction: SAINT (Bruker, 1998); program(s) used to solve structure: SIR92 (Altomare *et al.*, 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and CAMERON (Watkin *et al.*, 1993); software used to prepare material for publication: PLATON (Spek, 2003).

We thank Professor T. N. Guru Row, Department of Science and Technology, India, for data collection on the CCD facility set up under the IRHPA–DST program and Bangalore University. One of the authors (Vasu) thanks Vivekananda Degree College for support.

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